

# Chapter 2: Biological molecules

*AS Biology*

# General learning outcomes:

- **Describe** how large biological molecules are made from smaller molecules
- **Describe** the structure and function of carbohydrates, lipids and proteins

# Biological molecules intro

- The study of biological molecules forms an important branch of biology known as **molecular biology**
  - Molecular biology is closely linked with **biochemistry**, which looks at the chemical reactions of biological molecules
  - The sum total of all the biochemical reactions in the body is known as **metabolism**
  - The structure of molecules are closely related to

# Building blocks of life

- 4 most common elements in living organisms are, in order of abundance, hydrogen, carbon, oxygen and nitrogen
  - They account for more than 99% of the atoms found in all living things
  - Carbon atoms can join together to form long chains or ring structures
    - Can be thought of as the basic skeletons of *organic molecules* to which groups of other atoms are attached
      - *Organic molecules always contain carbon and hydrogen*

Figure 2.3, page 30

# Monomers, polymers and macromolecules

- **Macromolecule** means giant molecule
  - 3 types of macromolecules in living organisms:
    - polysaccharides, proteins (polypeptides) and nucleic acids (polynucleotides)
      - ‘poly’ means many; the above molecules are **polymers**, meaning that they are made of many repeating subunits that are similar or identical to each other

# Monomers, polymers and macromolecules

- The subunits are referred to as **monomers**
  - They're joined together like beads on a string
- Making polymers is relatively simple because the same reaction is repeated many times
- Natural examples of polymers are cellulose and rubber
- Industrially produced polymers include: polyester, polythene, PVC and nylon
  - These are all made up of carbon-based monomers and contain thousands of carbon atoms joined end to end

# Carbohydrates

- All carbohydrates contain the elements carbon, hydrogen and oxygen.
- ‘hydrate’ refers to water
  - Hydrogen and oxygen atoms are present in the ratio of 2:1 in carbohydrates, as they are in water
- The **general formula** for a carbohydrate:  $C_x(H_2O)_y$ .
- Three main groups of carbohydrates: monosaccharides, disaccharides and polysaccharides
  - ‘saccharide’ refers to a sugar or sweet substance



# Monosaccharides

- Monosaccharides are **sugars**
  - Sugars dissolve easily in water to form sweet-tasting solutions
  - General formula:  $(\text{CH}_2\text{O})_n$
  - Consist of a **single** sugar molecule; 'mono' means one

# Monosaccharides

- Main types of monosaccharides classified according to the number of carbon atoms in each molecule:
  - **Trioses (3C)**
  - **Pentoses (5C)**
    - Common examples: ribose and deoxyribose
  - **Hexoses (6C)**
    - Common examples: glucose, fructose and galactose
- Names of all sugars end with **-ose**

# Monosaccharides

- Molecular and structural formula
  - Formula (aka **molecular formula**) for a hexose:  
$$\text{C}_6\text{H}_{12}\text{O}_6$$
  - **Structural formula** shows the arrangements of the atoms in a molecule

# QUESTION

The formula for a hexose is  $C_6H_{12}O_6$ , or  $(CH_2O)$

6. What would be the formula of:

- a. a triose?
- b. a pentose?

Figure 2.3, page 30

# Ring structures

- Pentoses and hexoses have long enough chains of carbon atoms that they can close up on themselves and form more stable ring structures

Figure 2.4, page 30

# Roles of monosaccharides in living organisms

- 2 Major functions:
  - Commonly used as a source of energy in respiration due to the larger number of carbon-hydrogen bonds
    - C-H bonds are broken to release energy which is transferred to help make ATP (adenosine triphosphate) from ADP (adenosine diphosphate) and phosphate
  - Most important in energy metabolism is glucose



# Roles of monosaccharides in living organisms

- Important as building blocks for larger molecules
  - Example: glucose is used to make the polysaccharides starch, glycogen and cellulose
  - Example: ribose (5C) is one of the molecules used to make RNA (ribonucleic acid) and ATP
  - Example: Deoxyribose (5C) is one of the molecules used to make DNA